QAPTEC0023



Product Name: Nickel Metal Hydride Battery
Chemical Systems: Nickel Metal Hydride

Designed for Recharge: Yes

SECTION I - MANUFACTURER INFORMATION

Yuasa Battery Sales (UK) Ltd.

Unit 13, Hunts Rise,

South Marston Park, Swindon

SN3 4TG

Telephone Number for Information: +44 (0) 1793-833555

SECTION II - HAZARDOUS INGREDIENTS

IMPORTANT NOTE:

The battery should not be opened or burned. Exposure to the ingredients contained within or their combustion products could be harmful.

A) The content of elements are based on homogeneous materials level of NiMH battery:

Element	Lead	Cadmium	Hexavalent Chromium (Cr ₆₊)	Mercury	Polybrominated Biphenyls (PBBs)	Polybrominated Diphenyls Ethers(PBDEs)
% W. t.	< 0.1	< 0.01	< 0.1	<0.1	<0.1	<0.1

B) The content of elements are based on total weight of NiMH battery:

Nickel as nickel hydroxide nickel oxide nickel powder Potassium Hydroxide < 5 Cobalt	MATERIAL OR INGREDIENTS	% W. t.
nickel oxide nickel powder Potassium Hydroxide Cobalt as cobalt metal cobalt oxide cobalt hydroxide Sodium Hydroxide Zinc as zinc metal zinc oxide zinc hydroxide Mercury Lead as lead metal lead oxide	Nickel	30-50
nickel powder Potassium Hydroxide <5 Cobalt	as nickel hydroxide	
Potassium Hydroxide < 5 Cobalt as cobalt metal cobalt oxide cobalt hydroxide Sodium Hydroxide < 3 Zinc as zinc metal zinc oxide zinc hydroxide Mercury 0-0.0005 Lead as lead metal lead oxide	nickel oxide	
Cobalt as cobalt metal cobalt oxide cobalt hydroxide Sodium Hydroxide Zinc as zinc metal zinc oxide zinc hydroxide Mercury Lead as lead metal lead oxide 2.5-6.0 2.5-6.0 2.5-6.0 0-0.004	nickel powder	
as cobalt metal cobalt oxide cobalt hydroxide Sodium Hydroxide Zinc as zinc metal zinc oxide zinc hydroxide Mercury Defouting the state of the s	Potassium Hydroxide	< 5
cobalt oxide cobalt hydroxide Sodium Hydroxide Zinc as zinc metal zinc oxide zinc hydroxide Mercury Defourting the properties of the p	Cobalt	2.5-6.0
cobalt hydroxide Sodium Hydroxide Zinc	as cobalt metal	
Sodium Hydroxide < 3 Zinc	cobalt oxide	
Zinc as zinc metal zinc oxide zinc hydroxide Mercury Lead as lead metal lead oxide	cobalt hydroxide	
as zinc metal zinc oxide zinc hydroxide Mercury 10-0.0005 Lead as lead metal lead oxide 0-0.004	Sodium Hydroxide	< 3
zinc oxide zinc hydroxide Mercury 10-0.0005 Lead as lead metal lead oxide 0-0.004	Zinc	< 3
zinc hydroxide Mercury 0-0.0005 Lead 0-0.004 as lead metal lead oxide	as zinc metal	
Mercury 0-0.0005 Lead 0-0.004 as lead metal lead oxide	zinc oxide	
Lead 0-0.004 as lead metal lead oxide	zinc hydroxide	
as lead metal lead oxide	Mercury	0-0.0005
lead oxide	Lead	0-0.004
	as lead metal	
Cadmium < 0.002	lead oxide	
Caumum \Quad \Qu	Cadmium	<0.002
as cadmium metal	as cadmium metal	
cadmium oxide	cadmium oxide	
cadmium hydroxide		
Hexavalent Chromium (Cr ⁶⁺) 0-0.0005	Hexavalent Chromium (Cr ⁶⁺)	0-0.0005
Polybrominated Biphenyls (PBBs) Nil	Polybrominated Biphenyls (PBBs)	Nil
Polybrominated Diphenyls Ethers(PBDEs) Nil	Polybrominated Diphenyls Ethers(PBDEs)	Nil

YBSUK reserves the right to alter or amend the design, model and specification without prior notice.



OAPTEC0023

SECTION III - PHYSICAL / CHEMICAL CHARACTERISTICS

Boiling Point	Specific Gravity (H2O=1)		
N.A.	N.A.		
Vapor Pressure (mm Hg)	Melting Point		
N.A.	N.A.		
Vapor Density (AIR=1)	Evaporation Rate (Butyl Acetate)		
N.A.	N.A.		
Solubility in Water			
N.A.			
Appearance and Odor			
Cylindrical Shape, odorless			

SECTION IV - REACTIVITY DATA

Stability	Unstable		Conditions to Avoid		
	Stable	X			
Incompatibility (Materials to Avoid)					
Hazardous Decomposition or Byproducts					
Hazardous Polymerization	May Occur		Conditions to Avoid		
	Will Not Occur	X			

SECTION V - FIRE AND EXPLOSION HAZARD DATA

If fire or explosion occurs when batteries are on charge shut off power to charger.

In case of fire where nickel metal hydride batteries are present, apply a smothering agent such as METL-X, sand, dry ground dolomite, or soda ash, or flood the area with water. A smothering agent will extinguish burning nickel metal hydride batteries. Water may not extinguish burning batteries but will cool the adjacent batteries and control the spread of fire. Burning batteries will burn themselves out. Virtually all fires involving nickel metal hydride batteries can be controlled with water. However, when water is used hydrogen gas may evolve. In a confined space hydrogen gas can form an explosive mixture. In this situation smothering agents are recommended.

Fire fighters should wear self-contained breathing apparatus. Burning nickel metal hydride batteries can produce toxic fumes including oxides of nickel, cobalt, aluminum, manganese, lanthanum, cerium, neodymium, and praseodymium.

QAPTEC0023



SECTION VI - HEALTH HAZARD DATA

Under normal conditions of use, the battery is hermetically sealed.

Ingestion:

Swallowing a battery can be harmful.

Contents of an open battery can cause serious chemical burns of mouth, esophagus, and gastrointestinal tract.

Inhalation:

Contents of an open battery can cause respiratory irritation. Hypersensitivity to nickel can cause allergic pulmonary asthma. Provide fresh air and seek medical attention.

Skin Contact:

Contents of an open battery can cause skin irritation and/or chemical burns. Nickel, nickel compounds, cobalt, and cobalt compounds can cause skin sensitization and an allergic contact dermatitis. Remove contaminated clothing and wash skin with soap and water. If a chemical burn occurs or if irritation persists, seek medical attention.

Eve Contact:

Contents of an open battery can cause severe irritation and chemical burns. Immediately flush eyes thoroughly with water for at least 15 minutes, lifting upper and lower lids, until no evidence of the chemical remains. Seek medical attention.

SECTION VII- PRECAUTIONS FOR SAFE HANDLING AND USE

Storage:

Store in a cool, well ventilated area. Elevated temperatures can result in shortened battery life.

Mechanical Containment:

Never seal or encapsulate nickel metal hydride batteries.

Do not obstruct safety release vents on batteries. Encapsulation (potting) of batteries will not allow cell venting and can cause high pressure rupture.

Handling:

Accidental short circuit for a few seconds will not seriously affect the battery. However, this battery is capable of delivering very high short circuit currents. Prolonged short circuits will cause high cell temperatures which can cause skin burns. Sources of short circuits include jumbled batteries in bulk containers, metal jewelry, and metal covered tables or metal belts used for assembly of batteries into devices.

Do not open battery. The negative electrode material may be pyrophoric. Should an individual cell from a battery become disassembled, spontaneous combustion of the negative electrode is possible. This is much more likely to happen if the electrode is removed from its metal container. There can be a delay between exposure to air and spontaneous combustion.

YBSUK reserves the right to alter or amend the design, model and specification without prior notice.

CC0023 YUASA

QAPTEC0023

Charging:

This battery is made to be charged many times. Because it gradually loses its charge over a few months, it is good practice to charge battery before use. Use recommended charger. Improper charging can cause heat damage or even high pressure rupture. Observe proper charging polarity.

SECTION VIII-ACCIDENTAL RELEASE OR SPILLAGE

Steps to be taken in case material is released or spilled.

Batteries that are leakage should be handled with rubber gloves. Avoid direct contact with electrolyte. Wear protective clothing and a positive pressure Self-Contained Breathing Apparatus (SCBA).

SECTION IX-SPECIAL PROTECTION INFORMATION

Ventilation Requirements:

Not necessary under normal conditions.

Respiratory Protection:

Not necessary under normal conditions.

Eye Protection:

Not necessary under normal conditions. Wear safety glasses with side shields if handling an open or leaking battery.

Gloves:

Not necessary under normal conditions. Use neoprene or natural rubber gloves if handling an open or leaking battery.

Open Battery Storage:

Battery should not be opened. Should a cell become disassembled, the electrode should be stored in a fireproof cabinet, away from combustibles.

Keep batteries between -20°C and 35°C for prolong storage.

When the cells are closed to fully charged, the storage temperature should be between -20°C and 30°C and should be controlled at 10-20°C during transportation and packed with efficient air ventilation.



QAPTEC0023

SECTION X - EXPOSURE CONTROLS / PERSON PROTECTION

Occupational Exposure Limits: LTEP		STEP				
N.A.		N.A.				
Respiratory Pro	Respiratory Protection (Specify Type)					
	N.A.					
Ventilation	Local Exhausts	Special				
	N.A.	N.A.				
	Mechanical (General)	Other				
	N.A.	N.A.				
Protective Gloves		Eye Protection				
	N.A.	N.A.				
Other Protective Clothing or Equipment						
N.A.						
Work / Hygienic Practices						
N.A.						

SECTION XI - ECOLOGICAL INFORMATION

N.A.

SECTION XII - DISPOSAL METHOD

Dispose of batteries according to government regulations.

SECTION XIII - TRANSPORTATION INFORMATION

YBSUK here by certify that the above NiMH cylindrical batteries are non-dangerous and non-hazardous materials for air transport in any nature. The consignment is fully described by the proper shipping name and packed to prevent short circuit and from movement that could lead to short-circuiting, marked and in proper conditions for carriage conditions by air.

We here certify that the above batteries are not classified as dangerous under the current edition of IATA Dangerous Goods Regulation 56th edition and comply to the provision of A199 and all applicable and government regulations.

SECTION XIV - REGULATORY INFORMATION

Special requirement be according to the local regulatories.

SECTION XV - OTHER INFORMATION

The data in this Material Safety Data Sheet relates only to the specific material designated herein.

YBSUK reserves the right to alter or amend the design, model and specification without prior notice.

QAPTEC0023



SECTION XVI - MEASURES FOR FIRE EXTINCTION

In case of fire, it is permissible to use any class of extinguishing medium on these batteries or their packing material. Cool exterior of batteries if exposed to fire to prevent rupture.

Fire fighters should wear self-contained breathing apparatus.

Page 6 of 6 DCR00046 Issue 3 : 30-07-2015